

organic compounds



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1-(5,5-Dioxido-10*H*-phenothiazin-10-yl)ethanoneM. S. Siddegowda,^a Jerry P. Jasinski,^{b*} James A. Golen^b and H. S. Yathirajan^a^aDepartment of Studies in Chemistry, University of Mysore, Manasagangotri, Mysore 570 006, India, and ^bDepartment of Chemistry, Keene State College, 229 Main Street, Keene, NH 03435-2001, USACorrespondence e-mail: jjasinski@keene.edu

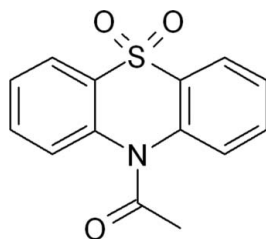
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Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.036; wR factor = 0.103; data-to-parameter ratio = 15.0.

In the title compound, $\text{C}_{14}\text{H}_{11}\text{NO}_3\text{S}$, the six-membered thiazine ring fused to two benzene rings adopts a distorted boat conformation. The dihedral angle between the mean planes of the two benzene rings is $45.8(1)^\circ$. The crystal packing is stabilized by weak intermolecular $\text{C}-\text{H}\cdots\text{O}$ interactions.

Related literature

For synthetic dyes and electroluminescent materials containing phenothiazine, see: Miller *et al.* (1999). For antipsychotic drugs, see: Wermuth *et al.* (2003). For applications of phenothiazine derivatives in medicine, see: Wang *et al.* (2008). For their antitumor activity, see: Lam *et al.* (2001). For related structures, see: Harrison *et al.* (2007); Jasinski *et al.* (2011). For standard bond lengths, see Allen *et al.* (1987). For puckering parameters, see: Cremer & Pople (1975).



Experimental

Crystal data

 $\text{C}_{14}\text{H}_{11}\text{NO}_3\text{S}$
 $M_r = 273.30$
 Monoclinic, $P2_1/c$
 $a = 12.5715$ (6) Å

 $b = 8.7648$ (4) Å
 $c = 11.5828$ (5) Å
 $\beta = 92.142$ (4) $^\circ$
 $V = 1275.38$ (10) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.26$ mm⁻¹
 $T = 173$ K
 $0.35 \times 0.15 \times 0.15$ mm

Data collection

 Oxford Diffraction Xcalibur Eos
 Gemini diffractometer
 Absorption correction: multi-scan
 (*CrysAlis RED*; Oxford
 Diffraction, 2010)
 $T_{\min} = 0.916$, $T_{\max} = 0.963$

 5342 measured reflections
 2597 independent reflections
 2263 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.019$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.036$
 $wR(F^2) = 0.103$
 $S = 1.02$
 2597 reflections

 173 parameters
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.26$ e Å⁻³
 $\Delta\rho_{\min} = -0.40$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C2}-\text{H2B}\cdots\text{O2}^{\text{i}}$	0.95	2.50	3.246 (2)	135
$\text{C8}-\text{H8A}\cdots\text{O1}^{\text{ii}}$	0.95	2.54	3.376 (2)	147
$\text{C9}-\text{H9A}\cdots\text{O3}^{\text{ii}}$	0.95	2.56	3.322 (2)	137

Symmetry codes: (i) $-x + 1, -y + 1, -z + 2$; (ii) $x, -y + \frac{3}{2}, z - \frac{1}{2}$.

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2010); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: YK2011).

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1-(5,5-Dioxido-10*H*-phenothiazin-10-yl)ethanone

M. S. Siddegowda, Jerry P. Jasinski, James A. Golen and H. S. Yathirajan

S1. Comment

Phenothiazine is a well known heterocycle. The phenothiazine structure occurs in many synthetic dyes, electroluminescent materials (Miller *et al.*, 1999) and drugs, especially various antipsychotic drugs, e.g. chlorpromazine and antihistaminic drugs, e.g. promethazine (Wermuth, 2003). Recently, researchers have found some new applications for phenothiazine derivatives in medicine related to antitubercular (Wang *et al.*, 2008) and antitumor activities (Lam *et al.*, 2001). The title compound has been used in the synthesis of oxomemazine, an antihistamine and anticholinergic drug of the phenothiazine chemical class used for the treatment of coughs. The crystal structures of dioxopromethazinium picrate (Harrison *et al.*, 2007) and 1-(10*H*-phenothiazin-2-yl)ethanone (Jasinski *et al.*, 2011) have been reported. In view of the importance of phenothiazines, this paper reports the crystal structure of the title compound, C₁₄H₁₁NO₃S.

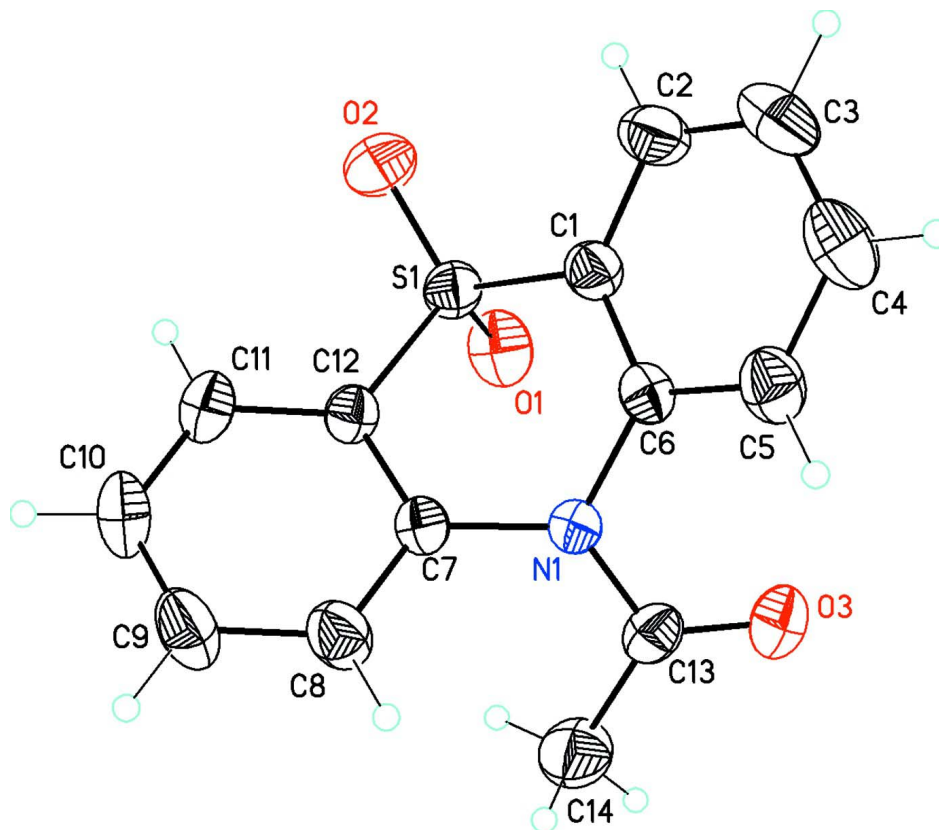
In the title compound the six-membered thiazine ring fused to two benzene rings adopts a distorted boat conformation. (Cremer & Pople, 1975) with puckering parameters Q , θ and ϕ of 0.6257 (12) Å, 95.85 (13)° and 180.29 (15)°, respectively (Fig. 1). For an ideal boat θ and ϕ have values of 90.0° and 180°. The dihedral angle between the mean planes of the two benzene rings is 45.8 (1)°. The ethanone group extends away from corner of the boat crease with a -169.76 (14)° (C6/N1/C13/C14) torsion angle. The SO₂ group extends away from the opposite corner of the boat crease with a 105.8 (14)° (C2/C1/S1/O1) torsion angle. Bond lengths are in normal ranges (Allen *et al.*, 1987). Crystal packing is stabilized by weak C—H...O (Table 1, Fig. 2) intermolecular interactions.

S2. Experimental

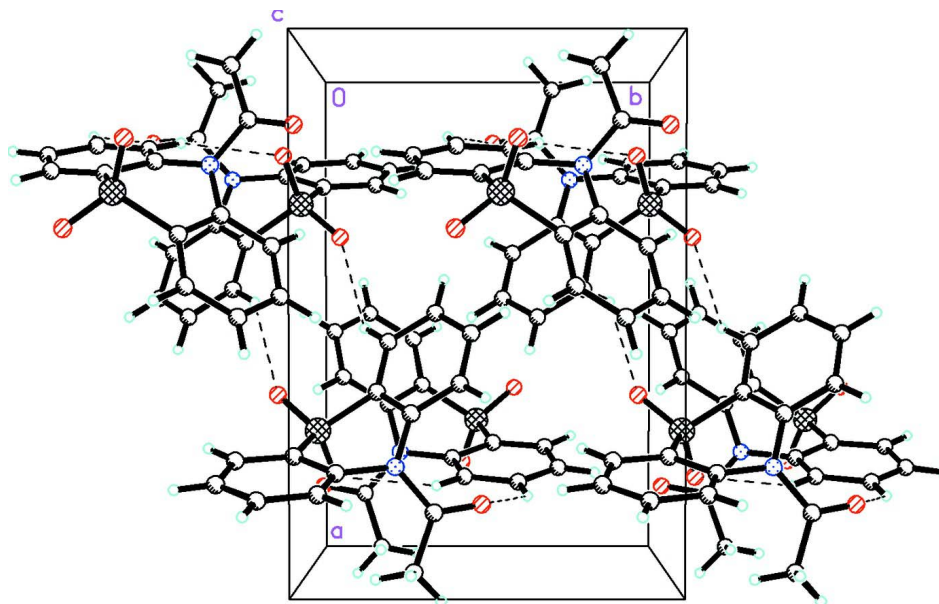
The title compound was obtained as a gift sample from RL Fine Chem, Bangalore, India. X-ray quality crystals were obtained by slow evaporation of solution of a 1:1 mixture of acetone:ethanol (m.p.: 495 K).

S3. Refinement

All of the H atoms were placed in their calculated positions and then refined using the riding model with Atom—H lengths of 0.95 Å (CH), or 0.98 Å (CH₃). Isotropic displacement parameters for these atoms were set to 1.19–1.21 (CH) or 1.49 (CH₃) times U_{eq} of the parent atom.

**Figure 1**

Molecular structure of the title compound showing the atom labeling scheme and 50% probability displacement ellipsoids.

**Figure 2**

Packing diagram of the title compound viewed down the *c* axis. Dashed lines indicate weak C—H...O intermolecular interactions.

1-(5,5-Dioxido-10*H*-phenothiazin-10-yl)ethanone*Crystal data*C₁₄H₁₁NO₃S $M_r = 273.30$ Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

 $a = 12.5715$ (6) Å $b = 8.7648$ (4) Å $c = 11.5828$ (5) Å $\beta = 92.142$ (4)° $V = 1275.38$ (10) Å³ $Z = 4$ $F(000) = 568$ $D_x = 1.423$ Mg m⁻³Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 3700 reflections

 $\theta = 3.3$ – 32.3 ° $\mu = 0.26$ mm⁻¹ $T = 173$ K

Block, colorless

 $0.35 \times 0.15 \times 0.15$ mm*Data collection*Oxford Diffraction Xcalibur Eos Gemini
diffractometer

Radiation source: Enhance (Mo) X-ray Source

Graphite monochromator

Detector resolution: 16.1500 pixels mm⁻¹ ω scans

Absorption correction: multi-scan

(CrysAlis RED; Oxford Diffraction, 2010)

 $T_{\min} = 0.916$, $T_{\max} = 0.963$

5342 measured reflections

2597 independent reflections

2263 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.019$ $\theta_{\max} = 26.4$ °, $\theta_{\min} = 3.3$ ° $h = -15 \rightarrow 15$ $k = -10 \rightarrow 10$ $l = -14 \rightarrow 13$ *Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.036$ $wR(F^2) = 0.103$ $S = 1.02$

2597 reflections

173 parameters

0 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0596P)^2 + 0.3216P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.26$ e Å⁻³ $\Delta\rho_{\min} = -0.40$ e Å⁻³*Special details*

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.27860 (3)	0.53431 (4)	0.88913 (3)	0.03458 (15)
O1	0.18673 (10)	0.57759 (15)	0.95043 (10)	0.0481 (3)
O2	0.34587 (11)	0.41839 (15)	0.93886 (12)	0.0552 (4)
O3	0.14803 (10)	0.98466 (14)	0.75069 (12)	0.0492 (3)

N1	0.22033 (10)	0.75636 (14)	0.71011 (11)	0.0301 (3)
C1	0.35385 (12)	0.69784 (18)	0.86054 (13)	0.0330 (3)
C2	0.44865 (13)	0.7260 (2)	0.92270 (15)	0.0441 (4)
H2B	0.4745	0.6564	0.9801	0.053*
C3	0.50395 (14)	0.8569 (2)	0.89904 (17)	0.0521 (5)
H3A	0.5685	0.8793	0.9410	0.063*
C4	0.46595 (15)	0.9559 (2)	0.81450 (18)	0.0523 (5)
H4A	0.5045	1.0467	0.7999	0.063*
C5	0.37288 (14)	0.9262 (2)	0.75021 (15)	0.0415 (4)
H5A	0.3487	0.9941	0.6909	0.050*
C6	0.31596 (11)	0.79526 (17)	0.77439 (13)	0.0307 (3)
C7	0.21083 (11)	0.60193 (17)	0.67173 (13)	0.0303 (3)
C8	0.18008 (14)	0.5650 (2)	0.55873 (15)	0.0413 (4)
H8A	0.1651	0.6434	0.5038	0.050*
C9	0.17142 (16)	0.4131 (2)	0.52666 (16)	0.0486 (5)
H9A	0.1488	0.3881	0.4498	0.058*
C10	0.19506 (15)	0.2974 (2)	0.60424 (16)	0.0470 (4)
H10A	0.1866	0.1939	0.5814	0.056*
C11	0.23104 (13)	0.33228 (18)	0.71510 (15)	0.0388 (4)
H11A	0.2510	0.2536	0.7680	0.047*
C12	0.23754 (12)	0.48386 (17)	0.74801 (13)	0.0301 (3)
C13	0.13555 (12)	0.85896 (18)	0.70914 (14)	0.0355 (4)
C14	0.03011 (14)	0.8073 (2)	0.65925 (19)	0.0527 (5)
H14A	−0.0254	0.8789	0.6816	0.079*
H14B	0.0140	0.7054	0.6886	0.079*
H14C	0.0326	0.8039	0.5748	0.079*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0430 (2)	0.0320 (2)	0.0286 (2)	0.00086 (16)	−0.00171 (16)	0.00387 (15)
O1	0.0580 (8)	0.0502 (7)	0.0371 (6)	−0.0057 (6)	0.0143 (6)	−0.0048 (6)
O2	0.0673 (9)	0.0431 (7)	0.0538 (8)	0.0057 (6)	−0.0169 (7)	0.0151 (6)
O3	0.0556 (8)	0.0313 (6)	0.0604 (8)	0.0070 (5)	−0.0022 (6)	−0.0060 (6)
N1	0.0334 (6)	0.0253 (6)	0.0311 (6)	−0.0007 (5)	−0.0050 (5)	0.0007 (5)
C1	0.0332 (7)	0.0339 (8)	0.0319 (8)	0.0001 (6)	−0.0005 (6)	−0.0033 (6)
C2	0.0391 (8)	0.0518 (10)	0.0406 (9)	0.0034 (8)	−0.0086 (7)	−0.0069 (8)
C3	0.0363 (9)	0.0687 (13)	0.0509 (11)	−0.0097 (9)	−0.0056 (8)	−0.0147 (10)
C4	0.0475 (10)	0.0565 (12)	0.0532 (11)	−0.0216 (9)	0.0068 (9)	−0.0083 (9)
C5	0.0473 (9)	0.0389 (9)	0.0385 (9)	−0.0100 (7)	0.0036 (7)	0.0000 (7)
C6	0.0322 (7)	0.0300 (7)	0.0299 (7)	−0.0021 (6)	0.0011 (6)	−0.0044 (6)
C7	0.0310 (7)	0.0268 (7)	0.0330 (8)	−0.0024 (6)	−0.0006 (6)	−0.0009 (6)
C8	0.0534 (10)	0.0380 (9)	0.0321 (8)	−0.0081 (8)	−0.0040 (7)	0.0014 (7)
C9	0.0667 (12)	0.0443 (10)	0.0346 (9)	−0.0121 (9)	−0.0004 (8)	−0.0097 (8)
C10	0.0620 (11)	0.0310 (9)	0.0485 (10)	−0.0071 (8)	0.0104 (9)	−0.0115 (8)
C11	0.0467 (9)	0.0269 (8)	0.0432 (9)	0.0013 (7)	0.0080 (7)	−0.0001 (7)
C12	0.0326 (7)	0.0280 (7)	0.0297 (7)	0.0006 (6)	0.0031 (6)	−0.0008 (6)
C13	0.0401 (8)	0.0311 (8)	0.0351 (8)	0.0023 (6)	0.0018 (6)	0.0060 (7)

C14	0.0363 (9)	0.0519 (11)	0.0697 (13)	0.0035 (8)	-0.0019 (9)	0.0010 (10)
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Geometric parameters (Å, °)

S1—O2	1.4292 (13)	C5—C6	1.387 (2)
S1—O1	1.4294 (13)	C5—H5A	0.9500
S1—C12	1.7524 (16)	C7—C8	1.389 (2)
S1—C1	1.7557 (16)	C7—C12	1.394 (2)
O3—C13	1.210 (2)	C8—C9	1.386 (2)
N1—C13	1.394 (2)	C8—H8A	0.9500
N1—C7	1.4283 (19)	C9—C10	1.380 (3)
N1—C6	1.4314 (19)	C9—H9A	0.9500
C1—C6	1.384 (2)	C10—C11	1.380 (2)
C1—C2	1.391 (2)	C10—H10A	0.9500
C2—C3	1.374 (3)	C11—C12	1.384 (2)
C2—H2B	0.9500	C11—H11A	0.9500
C3—C4	1.380 (3)	C13—C14	1.496 (2)
C3—H3A	0.9500	C14—H14A	0.9800
C4—C5	1.388 (3)	C14—H14B	0.9800
C4—H4A	0.9500	C14—H14C	0.9800
O2—S1—O1	117.73 (8)	C8—C7—C12	118.45 (14)
O2—S1—C12	110.23 (8)	C8—C7—N1	122.08 (14)
O1—S1—C12	108.36 (7)	C12—C7—N1	119.41 (13)
O2—S1—C1	109.97 (8)	C9—C8—C7	119.52 (16)
O1—S1—C1	109.16 (7)	C9—C8—H8A	120.2
C12—S1—C1	99.91 (7)	C7—C8—H8A	120.2
C13—N1—C7	123.62 (12)	C10—C9—C8	121.26 (16)
C13—N1—C6	118.52 (13)	C10—C9—H9A	119.4
C7—N1—C6	116.52 (12)	C8—C9—H9A	119.4
C6—C1—C2	121.92 (15)	C11—C10—C9	119.88 (16)
C6—C1—S1	117.79 (11)	C11—C10—H10A	120.1
C2—C1—S1	120.29 (13)	C9—C10—H10A	120.1
C3—C2—C1	118.35 (17)	C10—C11—C12	118.89 (16)
C3—C2—H2B	120.8	C10—C11—H11A	120.6
C1—C2—H2B	120.8	C12—C11—H11A	120.6
C2—C3—C4	120.13 (17)	C11—C12—C7	121.87 (15)
C2—C3—H3A	119.9	C11—C12—S1	120.77 (12)
C4—C3—H3A	119.9	C7—C12—S1	117.36 (11)
C3—C4—C5	121.69 (17)	O3—C13—N1	119.75 (15)
C3—C4—H4A	119.2	O3—C13—C14	121.93 (15)
C5—C4—H4A	119.2	N1—C13—C14	118.29 (14)
C6—C5—C4	118.54 (17)	C13—C14—H14A	109.5
C6—C5—H5A	120.7	C13—C14—H14B	109.5
C4—C5—H5A	120.7	H14A—C14—H14B	109.5
C1—C6—C5	119.33 (14)	C13—C14—H14C	109.5
C1—C6—N1	119.19 (13)	H14A—C14—H14C	109.5
C5—C6—N1	121.45 (14)	H14B—C14—H14C	109.5

O2—S1—C1—C6	154.78 (12)	C13—N1—C7—C12	−121.42 (16)
O1—S1—C1—C6	−74.65 (14)	C6—N1—C7—C12	45.13 (19)
C12—S1—C1—C6	38.88 (13)	C12—C7—C8—C9	3.4 (2)
O2—S1—C1—C2	−24.71 (16)	N1—C7—C8—C9	−179.49 (16)
O1—S1—C1—C2	105.85 (14)	C7—C8—C9—C10	−1.5 (3)
C12—S1—C1—C2	−140.62 (14)	C8—C9—C10—C11	−1.9 (3)
C6—C1—C2—C3	1.8 (3)	C9—C10—C11—C12	3.3 (3)
S1—C1—C2—C3	−178.75 (14)	C10—C11—C12—C7	−1.3 (2)
C1—C2—C3—C4	−0.8 (3)	C10—C11—C12—S1	177.93 (13)
C2—C3—C4—C5	−1.0 (3)	C8—C7—C12—C11	−2.0 (2)
C3—C4—C5—C6	1.7 (3)	N1—C7—C12—C11	−179.24 (14)
C2—C1—C6—C5	−1.1 (2)	C8—C7—C12—S1	178.73 (12)
S1—C1—C6—C5	179.43 (12)	N1—C7—C12—S1	1.50 (18)
C2—C1—C6—N1	177.23 (14)	O2—S1—C12—C11	26.68 (16)
S1—C1—C6—N1	−2.26 (19)	O1—S1—C12—C11	−103.47 (14)
C4—C5—C6—C1	−0.6 (2)	C1—S1—C12—C11	142.39 (13)
C4—C5—C6—N1	−178.90 (15)	O2—S1—C12—C7	−154.06 (12)
C13—N1—C6—C1	122.59 (15)	O1—S1—C12—C7	75.79 (13)
C7—N1—C6—C1	−44.68 (19)	C1—S1—C12—C7	−38.35 (13)
C13—N1—C6—C5	−59.1 (2)	C7—N1—C13—O3	174.72 (15)
C7—N1—C6—C5	133.59 (15)	C6—N1—C13—O3	8.4 (2)
C13—N1—C7—C8	61.5 (2)	C7—N1—C13—C14	−3.5 (2)
C6—N1—C7—C8	−131.99 (15)	C6—N1—C13—C14	−169.76 (14)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C2—H2B \cdots O2 ⁱ	0.95	2.50	3.246 (2)	135
C8—H8A \cdots O1 ⁱⁱ	0.95	2.54	3.376 (2)	147
C9—H9A \cdots O3 ⁱⁱ	0.95	2.56	3.322 (2)	137

Symmetry codes: (i) $-x+1, -y+1, -z+2$; (ii) $x, -y+3/2, z-1/2$.